# Effect of the Hinge Protein on the Heme Iron Site of Cytochrome $c_1^{\dagger}$

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Received November 23, 1988; Revised Manuscript Received December 27, 1988

ABSTRACT: X-ray absorption spectroscopic (XAS) studies on cytochrome  $c_1$  from beef heart mitochondria were conducted to identify the effect of the hinge protein [Kim, C. H., & King, T. E. (1983) J. Biol. Chem. 258, 13543-13551] on the structure of the heme site in cytochrome  $c_1$ . A comparison of XAS data of highly purified "one-band" and "two-band" cytochrome  $c_1$  [Kim, C. H., & King, T. E. (1987) Biochemistry 26, 1955-1961] demonstrates that the hinge protein exerts a rather pronounced effect on the heme environment of the cytochrome  $c_1$ : a conformational change occurs within a radius of approximately 5 Å from the heme iron in cytochrome  $c_1$  when the hinge protein is bound to cytochrome  $c_1$ . This result may be correlated with the previous observations that the structure and reactivity of cytochrome  $c_1$  are affected by the hinge protein [Kim, C. H., & King, T. E. (1987) Biochemistry 26, 1955-1961; Kim, C. H., Balny, C., & King, T. E. (1987) J. Biol. Chem. 262, 8103-8108].

Cytochrome  $c_1$  is an essential carrier in electron transfer from the cytochrome  $b-c_1$  complex to cytochrome c in the respiratory chain of mitochondria. Although structural properties of cytochrome  $c_1$ , including the amino acid sequence (Wakabayashi et al., 1982a; King, 1983; Kim & King, 1987), have been extensively studied, the molecular mechanism of electron transfer between cytochrome  $c_1$  and c is still not clearly understood because of a lack of detailed structural information.

Recently we have reported the role of the hinge protein (Kim & King, 1981, 1983; Wakabayashi et al., 1982b; Kim, 1987), which stabilizes the cytochrome  $c_1$ -c complex (Chiang et al., 1976; Kim & King, 1981), in electron-transfer reaction between cytochrome  $c_1$  and cytochrome c (Kim et al., 1987a,b). The effect of the hinge protein on some properties of cytochrome  $c_1$  has been also reported (Kim & King, 1987). In those reports, we proposed that the hinge protein may function as a regulator in the electron-transfer reaction between cytochromes  $c_1$  and c by stabilizing cytochrome  $c_1$  in a specific conformation.

In order to identify the effect of the hinge protein on the structure of cytochrome  $c_1$ , we performed X-ray absorption spectroscopic (XAS) studies of cytochrome  $c_1$  without the hinge protein ("one-band"  $c_1$ ) and of cytochrome  $c_1$  with the hinge protein ("two-band"  $c_1$ ). X-ray absorption spectroscopy is particularly suited to detect small structural differences such as subtle changes in heme stereochemistry (Shulman et al., 1978), as has been successfully demonstrated for cytochrome  $c_1$  (Labhardt & Yuen, 1979; Korszun et al., 1982). In this paper we present the first X-ray spectroscopic evidence for conformational changes on the heme iron site of cytochrome  $c_1$  when the hinge protein is bound to cytochrome  $c_1$ .

#### EXPERIMENTAL PROCEDURES

Enzyme Preparations. One-band and two-band cytochromes  $c_1$  were prepared by the method previously reported (Kim & King, 1987). Cytochromes  $c_1$  and c were oxidized with an excess amount of solid potassium ferricyanide and passed through a Sephadex G-25 column equilibrated with 50 mM phosphate buffer, pH 7.4. Cytochromes  $c_1$  (both twoband  $c_1$  and one-band  $c_1$ ) were then concentrated in an Amicon ultrafiltration cell with a PM-10 Diaflo ultrafiltration membrane or an Amicon Macrosolute concentrator (Minicon B15). All samples were stored at -80 °C prior to use.

X-ray Absorption Studies. X-ray absorption experiments were performed at room temperature at the National Biostructures PRT beam line X9-A of the National Synchrotron Light Source (NSLS) at Brookhaven National Laboratory. Cytochrome solutions were placed in a 1-mL Mylar window sample cell, and fluorescence XAS spectra were measured in the front-face fluorescence mode with a Stern-Heald-type ionization chamber with a MnO<sub>2</sub> filter and Soller slits for suppression of elastically scattered background radiation. Silicon-111 monochromator crystals were used, and the monochromator was detuned to minimize third harmonic contamination. Each sample was oriented such that the front face of the sample cell was at 45° with respect to both the direction of the X-ray beam and detector. Spectra were accumulated in 5-eV steps with 1.0-s collection time per step from 6.9 to 7.9 keV except for 2-eV steps between 6.96 and 7.19 keV.

X-ray absorption data were collected first for the oxidized cytochromes: one-band cytochrome  $c_1$ , 1-2 mM; two-band cytochrome  $c_1$ , 1.5-2.5 mM; horse heart cytochrome c, 2-4 mM, in 50 mM phosphate buffer, pH 7.4. After each EXAFS experiment a small amount for sample was withdrawn for optical spectral analysis to verify if radiation damage or photoreduction occurred in the sample during X-ray exposure. The rest of the sample was then reduced by adding an excess amount of solid sodium dithionite, and EXAFS experiments on the reduced cytochromes were performed.

## RESULTS AND DISCUSSION

Figure 1 shows the background-corrected fluorescence EXAFS spectra of both oxidized and reduced forms of horse

<sup>&</sup>lt;sup>†</sup>This work was supported by grants from the American Heart Association (861254) and the National Institutes of Health (RR01633, GM16767, and GM39612).

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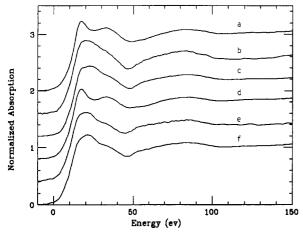


FIGURE 1: X-ray K-edge fluorescence excitation spectra of cytochromes  $c_1$  and c: (a-c) oxidized forms of two-band  $c_1$  (cytochrome  $c_1$  with the hinge protein), one-band  $c_1$  (cytochrome  $c_1$  without the hinge protein), and cytochrome c, respectively; (d-f) reduced forms of two-band  $c_1$ , one-band  $c_1$ , and cytochrome c, respectively. The preedge linear background has been subtracted, and the data are normalized to unit edge step. The spectra are offset for clarity. The energy zero is 7100 eV.

heart cytochrome c, one-band cytochrome  $c_1$ , and two-band cytochrome  $c_1$ . The spectra of oxidized and reduced horse heart cytochrome c agree well with those previously reported by Korszun et al. (1982). The spectrum of one-band cytochrome  $c_1$  is quite similar to that of the corresponding redox state of cytochrome c, although some small differences are evident. In contrast, the spectra of two-band cytochrome  $c_1$ are strikingly different from those of cytochrome c and oneband  $c_1$ . Specifically, double peaks immediately above the edge were observed in the spectrum of two-band  $c_1$ . However, this double peak structure cannot be a transient experimental artifact, because two separate experiments using two different enzyme preparations gave the same result. Furthermore, the data acquisition interval for the two-band  $c_1$  was interleaved with those for cytochrome c and one-band  $c_1$ , and the spectra of cytochrome c and one-band  $c_1$  have singly peaked principal maxima, although some fine structure is evident. Radiation damage is not a likely explanation, because no progressive changes in the spectra with time were observed, and the optical spectra appeared to be normal after X-ray irradiation. It is possible that the observed effect is due to the reaction of two-band cytochrome  $c_1$  with X-ray photoproducts, but this seems to be unlikely, since the heme iron site in two-band cytochrome  $c_1$  is less exposed than that in the one-band cytochrome  $c_1$  (Kim & King, 1987). One would expect that one-band cytochrome  $c_1$  would be more easily altered than two-band cytochrome  $c_1$ .

X-ray absorption spectra are sensitive to structural changes only within a radius of approximately 5 Å from the photo-absorbing iron atom because of the relatively short photo-electron mean free path (Stern, 1974) and the intrinsic lifetime broadening of the core hole. Therefore, it can be presumed that substantial conformational changes occur in the neighborhood of the iron atom when the hinge protein is bound to cytochrome  $c_1$ . The structural significance of this observation is not yet clear in detail, but possible explanations are (i) replacement of a first-shell ligand with another type; (ii) other strereochemical changes in the first coordination sphere, with retention of ligands; and (iii) stereochemical changes beyond first coordination spheres, i.e., changes in heme conformation.

The EXAFS spectra and corresponding Fourier transforms of cytochrome c and one-band cytochrome  $c_1$  can be super-

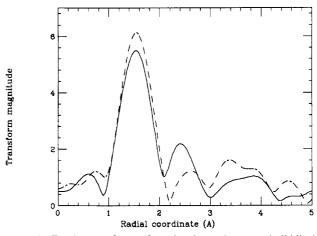


FIGURE 2: Fourier transforms of two-band cytochrome  $c_1$  (solid line) and cytochrome c (dashed line) EXAFS data. Both cytochromes are oxidized forms. The background subtracted data were  $k^3$  weighted, and the transform range is  $k = 3.0-8.8 \text{ Å}^{-1}$ . The magnitude of transformed spectra is in normalized arbitrary units.

imposed except for somewhat higher noise in the one-band cytochrome  $c_1$  spectrum (see Figure 1). Figure 2 shows the Fourier transforms of the EXAFS data of oxidized two-band cytochrome  $c_1$  and cytochrome c. The Fourier transform spectrum of two-band cytochrome  $c_1$  is compared with that of cytochrome c, because the quality of the Fourier transform of cytochrome c is better than that of one-band cytochrome c<sub>1</sub>. The background subtracted EXAFS data were weighted by  $k^3$ , and the transform range was  $k = 3.0-8.8 \text{ Å}^{-1}$ . The first-shell Fourier transform peaks of two-band cytochrome  $c_1$  and cytochrome c are very similar in modulus (Figure 2) and phase (not shown). In contrast, the second-shell transform peak of two-band cytochrome  $c_1$  is significantly different from that of cytochrome c, being aboout 90° out of phase on average, and the amplitude of the corresponding peak of twoband cytochrome  $c_1$  is slightly larger than that of cytochrome c. This shift in the average EXAFS phase over the data range may be caused by interferences between the atoms composing the second shell. The Fourier transforms of the data suggest that the first-shell environment of two-band cytochrome  $c_1$  is similar to that of cytochrome c, but there are significant structural differences beyond the first shell. Preliminary EXAFS analysis (Kim et al., 1987c) indicates that the average shell distances, coordination numbers, and Debye-Waller factors are only slightly different between two-band  $c_1$  and one-band  $c_1$ . From these results, it is conceivable that either the second or third coordination shell environments of the heme in cytochrome  $c_1$  may change when the hinge protein is bound to cytochrome  $c_1$ .

For all of the cytochromes, the positions of the edge (as judged by the maximum of the first derivative) are shifted slightly to lower energy upon reduction. Edge shifts are 1.8  $\pm$  0.3 eV for cytochrome c, 0.8  $\pm$  0.3 eV for one-band  $c_1$ , and 0.7  $\pm$  0.4 eV for two-band  $c_1$ . These observed edge shifts are comparable to the 1.5-eV shift reported by Korszun et al. (1982). The smaller shifts for cytochrome  $c_1$  are probably not caused by X-ray-induced reduction of the nominally oxidized samples, because the extent of reduction after EXAFS experiments was only 5–10% for both one-band and two-band cytochrome  $c_1$ .

Since the discovery of the hinge protein and its obligatory role for the formation of cytochrome  $c_1$ -c complex (Kim & King, 1981, 1983; Wakabayashi et al., 1982b; Kim, 1987), the function of the hinge protein in the mitochondria has remained to be investigated. Studies of structural and func-

tional properties of cytochrome  $c_1$  in the presence and absence of the hinge protein (Kim & King, 1987; Kim et al., 1984, 1987a) showed that the hinge protein may play a role for stabilizing the structure of cytochrome  $c_1$ . It appears that when the hinge protein is bound to the cytochrome  $c_1$ , the structure of cytochrome  $c_1$  is stabilized and somewhat protected from photoreduction, autoxidation, and other reactions (Kim & King, 1987). Mukai and Matsubara (1987) also demonstrated that the stable binding of the hinge protein to cytochrome  $c_1$  is necessary to protect cytochrome  $c_1$  from autoxidation.

Furthermore, we have recently found that the hinge protein seemed to play a certain role in electron-transfer reaction between cytochrome  $c_1$  and c, and the effect of the hinge protein is most probably mediated through a change of the conformation of cytochrome  $c_1$  (Kim et al., 1987a,b).

Although it is not clear at present if any explanation for structure-function relationships of cytochrome  $c_1$  with or without the hinge protein can be made from the current XAS results of cytochrome  $c_1$ , we are inclined to think that the observed effect of the hinge protein on the heme site of cytochrome  $c_1$  may be correlated with the observed function of the hinge protein in electron-transfer reaction between cytochrome  $c_1$  and c (Kim et al., 1987a,b). More detailed XAS studies will be conducted to substantiate our current hypothesis and, further, to characterize the structural changes induced in cytochrome  $c_1$  when the hinge protein is bound to cytochrome  $c_1$ .

#### **ACKNOWLEDGMENTS**

We acknowledge the valuable technical assistance of S. Chace and M. Seaman. We are grateful to Dr. J. C. Salerno for his valuable suggestions and discussion.

## REFERENCES

Chiang, Y.-L., Kaminski, L. S., & King, T. E. (1976) J. Biol. Chem. 251, 29-36.

- Kim, C. H. (1987) in Advances in Membrane Biochemistry and Bioenergetics (Kim, C. H., Tedeschi, H., Diwan, J. J., & Salerno, J. C., Eds.) pp 167-178, Plenum Press, New York.
- Kim, C. H., & King, T. E. (1981) Biochem. Biophys. Res. Commun. 101, 29-36.
- Kim, C. H., & King, T. E. (1983) J. Biol. Chem. 258, 13543-13551.
- Kim, C. H., & King, T. E. (1987) Biochemistry 26, 1955-1961.
- Kim, C. H., Balny, C., & King, T. E. (1984) *Proc. Natl. Acad. Sci. U.S.A.* 81, 2026–2029.
- Kim, C. H., Balny, C., & King, T. E. (1987a) J. Biol. Chem. 212, 8103-8108.
- Kim, C. H., Balny, C., & King, T. E. (1987b) in Cytochrome Systems: Molecular Biology and Bioenergetics (Papa, S., Chance, B., & Ernster, L., Eds.) pp 415-422, Plenum Press, New York.
- Kim, C. H., Bunker, G., Yencha, A. J., Chance, B., & King, T. E. (1987c) Biophys. J. 51, 492a.
- King, T. E. (1983) Adv. Enzymol. Relat. Areas Mol. Biol. 54, 267-366.
- Korszun, Z. R., Moffat, K., Frank, K., & Cusanovich, M. A. (1982) *Biochemistry 21*, 2253-2258.
- Labhardt, A., & Yuen, C. (1979) Nature 277, 150-151.
- Mukai, K., & Matsubara, H. (1987) in Advances in Membrane Biochemistry and Bioenergetics (Kim, C. H., Tedeschi, H., Diwan, J. J., & Salerno, J. C., Eds.) pp 179-184, Plenum Press, New York.
- Shulman, R. G., Eisenberger, P., & Kincaid, B. M. (1978)

  Annu. Rev. Biophys. Bioeng. 1, 559.
- Stern, E. A. (1974) Phys. Rev. B10, 3027-3037.
- Wakabayashi, S., Matsubara, H., Kim, C. H., & King, T. E. (1982a) J. Biol. Chem. 257, 9335-9344.
- Wakabayashi, S., Takeda, H., Matsubara, H., Kim, C. H., & King, T. E. (1982b) J. Biochem. (Tokyo) 91, 2077-2085.